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Key indicators

Single-crystal X-ray study
 T = 293 K
 Mean $\sigma(\text{C}-\text{C}) = 0.013 \text{ \AA}$
 R factor = 0.058
 wR factor = 0.185
 Data-to-parameter ratio = 16.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

Tris(*N*-ethyl-*N*-phenyldithiocarbamato-*S,S'*)cobalt(III)

The crystal and molecular structure of the title mononuclear Co^{III} complex, $[\text{Co}(\text{EtPhdte})_3]$ (EtPhdte is *N*-ethyl-*N*-phenyldithiocarbamate, $\text{C}_9\text{H}_{10}\text{NS}_2$), has been studied by single-crystal X-ray diffraction methods at 293 (2) K.

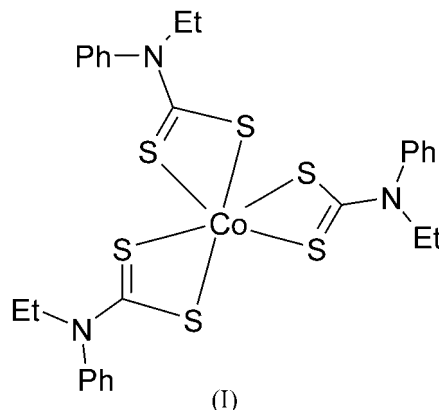
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Comment

The title compound, (I), was prepared in an exchange reaction between a chloroform solution of commercially manufactured vulcanization ultra-accelerator $[\text{Zn}(\text{EtPhdte})_3]$ (commercial designation Vulkacit P Extra; EtPhdte is *N*-ethyl-*N*-phenyldithiocarbamate) (Debnath & Basu, 1995) and an aqueous solution of cobalt(II) sulfate. The bond lengths in (I) are consistent with average values quoted in the usual sources (*International Tables for Crystallography*, 1992, Vol. C, Table 9.5.1.1).



Experimental

$[\text{Co}(\text{EtPhdte})_3]$ was prepared by the reaction of a chloroform solution of $[\text{Zn}(\text{EtPhdte})_2]$ and an aqueous solution of cobalt(II) sulfate at room temperature. The dark-brown product was recrystallized from chloroform.

Crystal data

$[\text{Co}(\text{C}_9\text{H}_{10}\text{NS}_2)_3]$
 $M_r = 647.83$
 Triclinic, $P\bar{1}$
 $a = 10.193 (2) \text{ \AA}$
 $b = 11.159 (2) \text{ \AA}$
 $c = 14.253 (3) \text{ \AA}$
 $\alpha = 105.14 (3)^\circ$
 $\beta = 103.09 (3)^\circ$
 $\gamma = 92.30 (3)^\circ$
 $V = 1515.7 (5) \text{ \AA}^3$

$Z = 2$
 $D_x = 1.419 \text{ Mg m}^{-3}$
 Mo $K\alpha$ radiation
 Cell parameters from 25 reflections
 $\theta = 3.8\text{--}9.1^\circ$
 $\mu = 1.00 \text{ mm}^{-1}$
 $T = 293 (2) \text{ K}$
 Prism, dark brown
 $0.6 \times 0.3 \times 0.3 \text{ mm}$

Data collection

Syntex $P2_1$ diffractometer
 θ - 2θ scans
 Absorption correction: ψ scan
 (North *et al.*, 1968)
 $T_{\min} = 0.544$, $T_{\max} = 0.741$
 6616 measured reflections
 6560 independent reflections
 2593 reflections with $I > 2\sigma(I)$

$R_{\text{int}} = 0.082$
 $\theta_{\text{max}} = 27.5^\circ$
 $h = -9 \rightarrow 1$
 $k = -14 \rightarrow 14$
 $l = -18 \rightarrow 18$
 3 standard reflections
 every 100 reflections
 intensity decay: 15%

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.058$
 $wR(F^2) = 0.185$
 $S = 0.86$
 5379 reflections
 334 parameters

H-atom parameters not refined
 $w = 1/[\sigma^2(F_o^2) + (0.1151P)^2]$
 where $P = (F_o^2 + 2F_c^2)/3$
 $(\Delta/\sigma)_{\text{max}} < 0.001$
 $\Delta\rho_{\text{max}} = 0.40 \text{ e } \text{\AA}^{-3}$
 $\Delta\rho_{\text{min}} = -0.34 \text{ e } \text{\AA}^{-3}$

Table 1

Selected geometric parameters (\AA , $^\circ$).

Co1—S2	2.256 (2)	Co1—S6	2.263 (2)
Co1—S4	2.255 (2)	Co1—S3	2.267 (2)
Co1—S5	2.259 (2)	Co1—S1	2.279 (2)
S2—Co1—S4	93.33 (8)	S5—Co1—S3	164.15 (7)
S2—Co1—S5	94.74 (7)	S6—Co1—S3	93.84 (7)
S4—Co1—S5	92.41 (8)	S2—Co1—S1	76.13 (7)
S2—Co1—S6	166.96 (7)	S4—Co1—S1	166.80 (7)
S4—Co1—S6	96.57 (8)	S5—Co1—S1	96.32 (8)
S5—Co1—S6	76.43 (7)	S6—Co1—S1	95.09 (8)
S2—Co1—S3	96.76 (7)	S3—Co1—S1	97.00 (8)
S4—Co1—S3	76.09 (7)		

Since the crystal diffracted very weakly, the completeness of the data is just 81.1% to 25° in θ .

Data collection: $P2_1$ *Diffractometer Control Software* (Syntex, 1973); cell refinement: $P2_1$ *Diffractometer Control Software*; data reduction: $XP21$ (Pavelčík, 1993); program(s) used to solve structure: $SHELXS86$ (Sheldrick, 1985); program(s) used to refine structure: $SHELXL97$ (Sheldrick, 1997); molecular graphics: $ORTEP$ (Johnson, 1965); software used to prepare material for publication: $SHELXL97$.

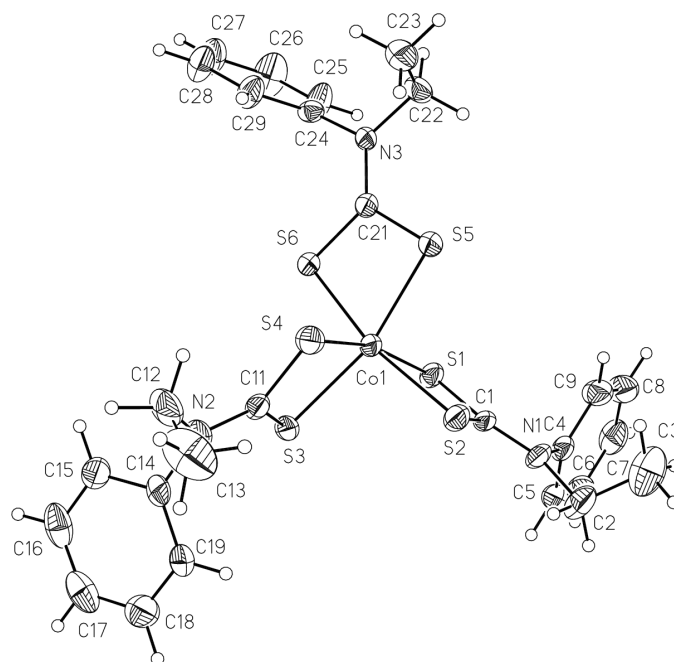


Figure 1

The structure of tris(*N*-ethyl-*N*-phenyldithiocarbamato-*S,S'*)cobalt(III) with displacement ellipsoids drawn at the 30% probability level.

References

- Debnath, S. C. & Basu, D. K. (1995). *Kautsch. Gummi Kunstst.* **48**, 111–116.
 Johnson, C. K. (1965). *ORTEP*. Report ORNL-3794. Oak Ridge National Laboratory, Tennessee, USA.
 North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). *Acta Cryst.* **A24**, 351–359.
 Pavelčík, F. (1993). *XP21*. Pharmaceutical Faculty, Comenius University, Bratislava, Slovakia.
 Sheldrick, G. M. (1985). *SHELXS86*. *Crystallographic Computing 3*, edited by G. M. Sheldrick, C. Krüger and R. Goddard, pp. 175–189. Oxford University Press.
 Sheldrick, G. M. (1997). *SHELXL97*. University of Göttingen, Germany.
 Syntex (1973). *P21, Diffractometer Control Software*. Syntex Analytical Instruments, Cupertino, California, USA.